RODER, Ivan; POROSZLAY, Borbala

Testing mechanical damages on cotton fibers. Magy textil 16 no. 2:49-54 F '64.

1. Textilipari Kutato Intezet.

Icrigation system for sugar beets in the steppe zerre Kuybyshew Province. Gidr. i mel. 16 no.11:3-7 N '64 (MJE. 18: )

1. Kuybyshevskaya sel'skokhozyaystvernaya opytanja c'antoire.

: USIR COUPERY : Soil Science, Tillage. Improvement. Ercsion. CATEGORY ARS. JEUR. : RZhulol., No. 3 1959, No. 10706 : Rorchagin, V. A. Porotikin, Ye. I. AUTFOR .: Kuybychev State Experiment Station INST. : Results of Experiments on the Study of New Methods TITLE of Soil Tillage. ORIG. PUB. : Byul. nauchno-tekhn. inform. Kuybyshevsk. (Bezerchuksk.) gos. s.-kh. cpytn at., 1957, 1, 9-11 : Deep subsurface tillage (without inversion) of bere fal-AYSTRACT low at Kuybahev Station secured better absorption of autumn-winter precipition, promoted destruction of weeds and increased the yield of rye grain in comparison with the usual lowing of bare fallow. The fall subsurface tillege (without inversion) increased the yield of grain grops by 1-3.5 centners/ha but on veed contaminated plants, it promoted a still greater contamination of the CATAD: 1/2 35

POROTNIKOV, A. A., KOSTYLEV, A. M., PELEVIN, V. N. (MOBCOW)

"On Some Equipment for High Temperature Plasma Studies and Some Experimental Data Concerning Magnetohydrodynamic Phenomena."

report presented at the First All-Union Congress on Theoretical and \*Applied Mechanics, Moscow, 27 Jan - 3 Feb 1960.

L 15726-63 EPR/ENT(1)/ENG(k)/EEC(b)-2/ES(w)-2/EDS ESD-3/IJP(C)/SSD

Ps-4/Pz-4/Pi-4/Po-4/Pab-4 WW/AT ACCESSION NR: AR3002665

S/0124/63/000/005/B017/B018

SOURCE: Rzh. Mekhanika, Abs. 5B86

AUTHOR: Kostylev, A.M.; Porotnikov, A. A.

TITLE: Plasma jet experiments in a magnetic field

CITED SOURCE: Sb. Vopr. magnitn. gidrodinamiki i dinamiki plazmy. v. 2. Riga, AN Latyssa, 1962, 477-488

TOPIC TAGS: plasmatron, jet, magnetic field, high temperature, Gvozdover formula, plasma arc, magnetic probe

TRANSLATION: Plasmatron experiments are described. To attain high plasma temperatures, an arc discharge was used in which the diameter of the arc channel is limited by a special disphragm with external or internal cooling by gas or liquid. The work was carried out on two devices. In one, with a power of 115 kw, the density of the current was 7100 smp/cm<sup>2</sup>, the intensity of the electric field was 115 volts/cm, the arc length 30-105 mm. In the other device, with a power of 4000 kv, the current density reached 74000 amp/cm<sup>2</sup>, the field 500 volts/cm, the

· 医克莱克氏 医软骨 经存货 "这一个,我们就是有数别的,我们就是这个人,我们就是这个人,我们就是这个人,我们就是这个人,我们就是这个人,我们就是这个人,我们	
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ACCESSION NR: AR3002665 The temperature was estimated through the power per unit volume 37000 kilowatt/cm <sup>3</sup> . The temperature was estimated through power per unit volume 37000 kilowatt/cm <sup>3</sup> . The temperature power to the formula of Gvozdover. Averaging along the electroconductivity according to the formula of Gvozdover. Averaging along the electroconductivity was 10 <sup>14</sup> the dismeter, the temperature proved to be equal to 280000, maximal temperature the dismeter, the temperature proved to be equal to 280000 and the conductivity was several kilowater.	
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DATE ACQ: 14Jun63	
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Card 2/2	
	. :

Intrustive rocks in the Sarbay deposit. Zap. LGI 47 no.2:114121 '64. (MIRA 18:3)

BURTSEVA, Z.A.; POROTOVA, G.A.

New data on alkali intrusions in the Turiy Peninsula. Mat. po geol. i pol. iskop. Sev.-Zap. RSFSR no.3:143-148 '62.

(MIRA 17:12)

TSIRUL'NIKOVA, I.Ya.; SHUSTOVA, L.Ye.; POROTOVA, G.A.

Deep-seated formations in the Pechenga structural zone according to geophysical data. Zap. IGI 46 no.2:14-16 (MIRA 17:6)

163.

POROTOVA, G.A.; SIPAKOVA, M.S.

Geological interpretation of the deep magnetic anomalies

of the Kola Peninsula. Zap. LGI 46 no.2:80-86 163.

(MIRA 17:6)

LEYKIN, Abram Yefimovich; POROTSKIY, Efroim Solomonovich; RODIN,
Boris Tosifovich; SAMOKHOTSKIY, A.I., inzh., retsenzent;
ZOL'NIKOVA, N.K., inzh., retsenzent; ROMADIN, K.P.,
kand. tekhn. nauk, red.

[Aircraft materials] Aviatsionnoe materialovedenie. Moskva, Mashinostroenie, 1964. 458 p. (MIRA 17:12)

6(4)

AUTHORS: Leonova, Z. M

Leonova, Z. M., Pass, M. I.,

sov/108-13-11-6/15

Porotskiy, F. Ya., Solov'yev, G. F.

TITLE:

Experience When Using Strong Oscillator Tubes in Impulse Operation (Opyt ispolizovaniya moshchnykh generatornykh lamp

v impul'snom rezhime)

PERIODICAL:

Radiotekhnika, 1958, Vol 13, Nr 11, pp 39-43 (USSR)

ABSTRACT:

The possibility of using strong oscillator tubes with active cathode, which are intended to be used for continuous operation, are investigated in pulsed apparatus. The preliminary tests, which were carried out by B. I. Polyakov, B. T. Zarubin, B. M. Gutner and K. N. Bulychev, gave positive results. On the strength of these results investigations of these tubes were carried out on a larger scale from 1955 to 1956. Work was carried out in two directions: 1) Testing of the tubes in static operation for the purpose of obtaining the entire family of static characteristics necessary for calculating the impulse-operation of the generator. 2) Control of dynamical operation for the purpose of checking the working of tubes in pulsed

Card 1/3

Experience When Using Strong Oscillator Tubes in Impulse Operation

SOV/108-13-11-6/15

operation and especially in ultrashortwave generator circuits to be used in practice. Possibilities of extending the frequency range and of considerably increasing the impulse power output are pointed out. The results obtained by investigations are discussed. The tests carried out showed that it is possible to use strong oscillator tubes for continuous work at low frequencies. The output values in the pulse obtained surpassed the nominal ones by a multiple (in the case of continuous operation). On the basis of the results obtained it may be concluded that it is possible to use strong oscillator tubes with thoriated cathodes in pulse generators at frequencies of 100 to 150 megacycles.

Z. I. Model', G. M. Drabkin, Z. M. Lifshits, and G. M. Moskovskaya advised the authors. A. I. Mermonshteyn, Engineer, and A. Ye. Karpova, Engineer, took part in the experiments.

Card 2/3

Experience When Using Strong Oscillator Tubes in SOV/108-13-11-6/15 Impulse Operation

There are 5 figures and 1 table.

SUBMITTED:

December 6, 1957

Card 3/3

POROTSKIY, Vg. M.

22505

Porotskiy, E. M. Issledovanie Vodouderzhivayushchey Sposobnosti
Tsementa. Trudy (Gos. Vsesoyuz Nauch - Issled. I I Proekt. In-T
Tsement Prom - Sti) VYP 10, 1949 S 3-59 Bibliogr S 57

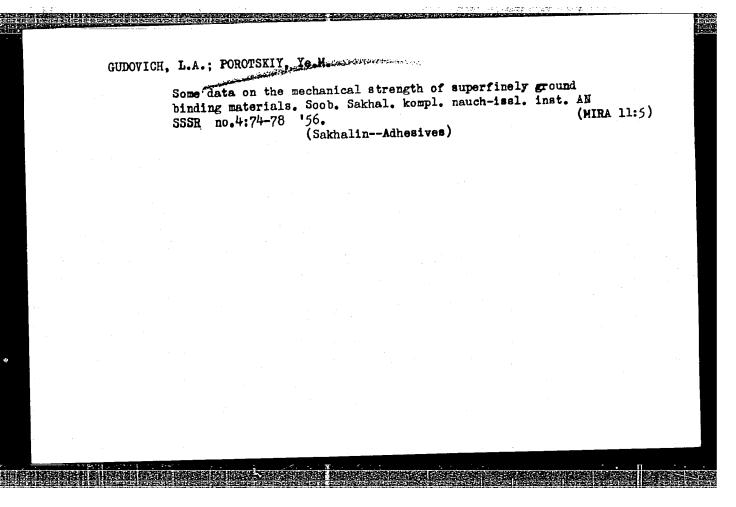
Letopis

So: Letopis' No 30, 1949

GUDOVICH, L.A.; FOROTSKIY, Ye.M.

Studying the properties of limestone-diatomaceous cement made with unslaked lime. Soob.Sakhal.kompl.nauch.-issl.inst.AN SSSR. no.2: (MIRA 14:4) 62-78 155.

(Cement)



USSR/Chemical Technology - Chemical Products and I-10 Their Applications - Silicates. Glass.

Ceramics. Binders.

: Ref Zhur - Khimiya, No 3, 1957, 9055 Abs Jour

Author

Gudovich, L.A., and Porotskiy, Ye.M.Sakhalin Branch of the Academy of Sciences Inst

: Investigation of the Properties of Limeston-Title

Diatomite Cement Prepared from Quicklime.

: Soobshch. Sakhalinsk. fil. AN SSSR, 1955, Orig Pub

No 2, 62-78

Physicochemical and acoustic investigations Abstract

and mechanical tests on the properties of Limestone-diatomite cement (LDC) have shown that the utilization of quicklime in the formulation of the cement leads to a considerable decrease in the setting time and

Card 1/3

PERCTSKIY, YE. M

I-10

USSR/Chemical Technology - Chemical Products and

Their Applications - Silicates. Glass.

Ceramics. Binders.

Ref Zhur - Khimiya, No 3, 1957, 9055 Abs Jour

> has an accelerating effect on the hardness of the cement compared to formulations in which limestone and pumice are used. The addition of alcoholic sulfite liquor (ASL) and of gypsum separately or, particularly, together considerably reduces the rate of hydration of the lime in the LDC. Gypsum and particularly ASL thin the cement paste and retard its setting, increasing the length of time during which the paste remains in a fluid state. The addition of gypsum markedly increases the compression strength of LDC as well as its modulus of rupture and its modulus of elasticity, especially during the initial hardening period;

Card 2/3

APPROVED FOR RELEASE 16/15/2010 E1ATROP86-00515R0015425201

POROTSKIY, Yu.I., gornyy inzh. elektromekhanik

Improved blocking of the disconnecting switch and plug-type clutch in the lead-in of the IGD-1 and IGD-2 cutter-leaders.

Ugol' 39 no.10:58 0 '64. (MIRA 17:12)

1. NIIOGR.

BREDIKHIN, A.M., prof.; POROTSKIY, Yu.I., kand.tekhn.nauk

"Mine lighting" by M.M. Fotiev, N.L. Sheviakov.
Reviewed by A.N. Bredikhin, IU.I. Porotskii.
Ugol' 37 no.9:61-62 S'62.

1. Zaveduyushchiy kafedroy gornoy elektrotekhniki
Sverdlovskogo gornogo instituta im. Vakhrusheva (for
Bredikhin).

(Mine lighting)

(Fotiev, M.M.)

(Sheviakov, N.L.)

POROTSKIY, Yu.-I.

At the Dnepropetrovsk-Mining Institute in Artem-Sergeyev from April-1939 to April 1947, the following dissertations were defended in connection with attaining the scholarly degree of Candidate of Technical Sciences (specializing in mining electrical engineering: Yu. I. Porotskiy on 30 December 1946 defended his dissertation on the subject "An investigation of the problems of transmitting electric power in mines where the deposits are near the surface".

The official opponents of this dissertation were Doctor of Technical Sciences Professor P. P. Pirotskiy, Candidate of Technical Sciences Docent R. Ya. Nayerov, and Docent T. A. Zanuzdannyy.

A technical-economic analysis was made of two methods of transmitting power in underground mining, in the main shaft of the mine and through special drill holes. It was shown that the second method is superior when deposits are located up to 250 meters below the surface.

SO: Elektrichestvo / Electricity /, No. 10, October 1947. Moscow.

RUMANIA / Soil Science. Gonosis and Geography of Soils. J-1

Abs Jour: Ref Zhur-Biol., No 8, 1958, 34319.

: Porovat, M., Nicolaescu-Plopsor, C., Spirescu, M.

Author

: Archeological Critoria for Establishment of Chron-Inst Titlo

ology in Paloo-Soil-Scienco.

Orig Pub: Commun. Acad. RPR, 1957, 7, No 3, 369-375.

Abstract: Morphology and conditions of stratification of ancient buried soils in Oltenia, in the vicinity of Plonitza, are described. The interred soil, rovoaled there during the excavation of a hillock, was attributed to the ancient black earth soils, formed - according to the authors - in the first period of the Bronze ago. Cited is

card 1/2

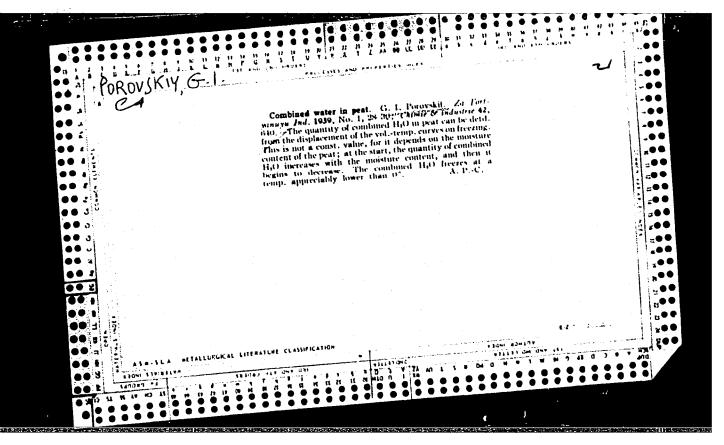
POROVIK, E. S. (Kharkov)

"Galvanomagnetic Effect and Properties of Conduction Electrons in Metals," a paper submitted at the International Conference on Physics of Magnetic Phenomena, Sverdlovsk, 23-31 May 56.

POROVNE, I.

Feeding high-frequency equipment. p. 18. TELEKOMUNIKACJE, Vol. 4, No. 3, July, 1955. Belgrad.

SOURCE: East European Accessions List (EEAL) Library of Congress, Vol. 4, No. 12, Dec. 1955.



USSR / Electricity

G

Abs Jour

: Ref Zhur - Fizika, No 4, 1957, No 9642

Author

: Balygin, I.E., Porovskiy, K.S.

Inst

: Not give

Title

: Aging of Insulation of Ceramics at High Temperature.

Orig Pub

: Zh. Tekhn. fiziki, 1956, 26, No 8, 1714-1722

Abstract

: A study was made of the processes of irreversible time change of electric properties of insulation of ceramic insulating materials at constant voltage. Tests were made on "ultra-porcelain" (UF-46), "radio-porcelain," steatite, (B-17), and spinel (sh-15). It is shown that the "ultra-porcalain" can bardly be used at 3800 even at an applied field intensity of 100 -- 200 volts/mm. In prolonged tests, there were noticed in the specimens formation of layers of greatly differing colors, this being due to the influence of the

Card

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USSR / Electricity

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Abs Jour

: Ref Zhur - Fizika, No 4, 1957, No 9642

Abstract

: electrolytic processes. A gradual destruction of silver electrodes and a certain deterioration of the dielectric properties (increase of tan 5 and reduction on resistance) were noted. A spectral analysis of various layers of the specimen disclosed and permitted a study of migration of metallic ions, entering into the composition of the ceramic and also of the ions of the silver elec trode and of the ions of bismuth and lead, which are contained in the paste used to coat the electrodes, which pass through the entire thickness of the specimen from the anode to the cathode. The electrolytical processes in "radio-porcelain" are so intense at 380°, that prolonged and reliable operation of the specimens is hardly possible even at field intensities of tens of volts per millimeter. Specimens of steatite can be operated for a long time at 380° with an intensity of 0.4 kv/mm.

Card : 2/3

USSR / Electricity

G

Abs Jour

: Ref Zhur - Fizika, No 4, 1957, No 9642

Abstract

: Specimens made of spinel were subjected to aging at 280° to a lesser degree than all the other tested materials. They can operate reliably for a long time at 380° at U = 0.8 kv/mm. The higher temperature indurance of steatite and spinel relative to the aging of insulation in prolonged action of dc voltage is due to the smaller contents of alkalimetal oxides. Being weakly attached to the ceramic, the alkali ions shift easily and cause a damage to the silver layer of the electrode and irreversible changes in the structure of the ceramic.

Card

: 3/3

AUTHORS:

Balygin, I. Ye., Porovskiy, K. S.

sov/57-58-8-11/37

TITLE:

On the Temperature Dependence of the Electric Strength of

Fluoroplast (O temperaturnoy zavisimosti elektricheskoy

prochnosti ftoroplasta)

PERIODICAL: Zhurnal tekhnicheskoy fiziki, 1958 Nr 8, pp. 1679 - 1683 (Nggu)

ABSTRACT:

Fluoroplast -4 (FP-4) is a polymer of polytetra-fluoro

ethylene (CF2 = CF2). In this paper data concerning the

temperature dependence of the breakdown voltage in fluoroplast

at a constant, at an alternating (50 c) and at a high-frequency voltage are given. Special electrodes mounted on holders of micalex (mikaleks) and a furnace for the electrodes was constructed for the experiments. As can be seen from the curves obtained the breakdown voltage of the film is hardly dependent upon temperature. When the number of layers in the film is increased the breakdown voltage also rises. This tends to show that the probability of a coincidence of the weak spots in the insulation is reduced when the layers are superimposed.

Card 1/3

The most pronounced drop of the breakdown voltage occurs

On the Temperature Dependence of the Electric Strength of Fluoroplast

507/57-58-8-11/37

above 200°C. This rule is only broken by the breakdown voltage  $E_{\text{breakdown}} = f(t)$  of one single layer, as in this case the dependence is linear. Formulae permitting to compute the breakdown voltage in fluoroplast in the range from 25 - 300°C are presented. For the sake of comparison the experimental data are given. A limit of the breakdown voltage is reached with a number of 8-9 layers. Now the curves showing the function of the breakdown voltage versus the total thickness of the packet at the respective temperatures are given. It was found that fluoroplast films heated to 200°C and above are cracked in cooling down. The breakdown voltage is dependent upon the electrode surface S when the number of layers is small. In order to arrive at definite conceptions concerning this phenomenon films were disrupted with electrodes. from aluminum foils, the films having a thickness of  $7-8\mu$ . = f(s) actually It appeared that such a function Ebreakdown exists. The breakdown voltage in fluoroplast decreases at high frequency. Breakdown experiments were conducted in various liquids. These experiments substantiated the assumptions

Card 2/3

On the Temperature Dependence of the Electric Strength of Fluoroplast

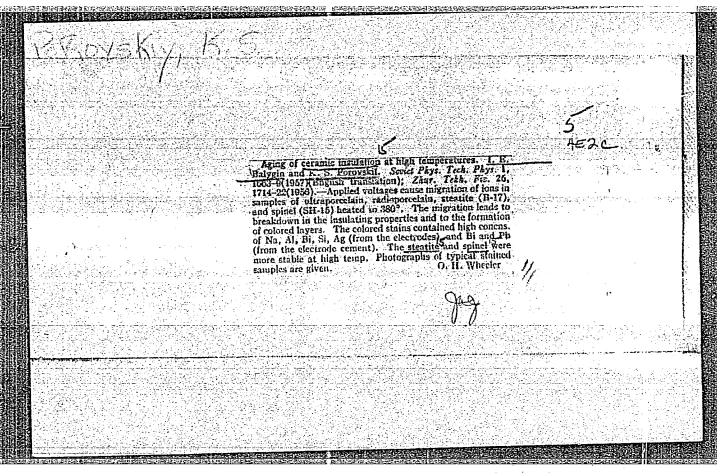
sov/57-58-6-11/37

concerning the development of breakdowns in the microfissures. When the number of layers is greater than three the breakdown voltage remains almost constant. The curves obtained show that the breakdown voltage is markedly dependent upon temperature at a constant voltage and at an alternating voltage of 50 c. There are 7 figures, 2 tables, and 8 references, 4 of which are Soviet.

SUBMITTED:

July 11, 1957

Card 3/3



AUTHOR: TITLE:

PA - 2541 BALYGIN, I.E., and POROVSKIY, K.S. Influence of Electrode Constituent on Ceramic Dielectrics Isolation Agening. (Vliyaniye metalla elektrodov na stareniye

izolyatsii keramicheskikh dielektrikov, Russian)

PERIODICAL

Zhurnal Tekhn. Fiz, 1957, Vol 27, Nr 3, pp 513 - 515 (U.S.S.R.) Received: 4 / 1957

ABSTRACT:

A short report concerning the results of endurance tests of models of the ceramic mass KW-1 (consisting chiefly of Al203 and SiO2 with small additions of BaO, CaO, SrO etc). and the statie-mass STs-4 (chiefly of SiO2 and MgO with small additions of BaO, Al203, Zn/O etc). The dielectric transmissivity at 1 MHz frequency is in the first case 7.2, in the second 6.7. The tests were carried out at a temperature of from 390 - 4000 C. The electrodes were applied by burning in silver at 800° C and Platinum at 900° C. In the case offmost samples electrodes of different metals were applied. Results are shown in a table. It may be taken for granted that the main

part of the aging of the insulation and the deterioration of its electric properties in the case of some ceramic insulators is due to silver penetrating into the thickness of the ceramic. In comparison platinum silver has a higher diffusion-rate with respect to some ceramic-sorts. The irrever-

Card 1/2

L 23800-66 EWT(m)/EWP(t) IJP(c) JD/JG ACC NR. AP6007251 (A) UR/0363/66/002/002/0275/0280 AUTHOR: Komissarova, L.N.; Po rovskiy, V.I.; Shaplygin, I.S. ORG: Moscow State University im. M.V. Lomonosov. Department of Chemistry (Moskovskiy gosudarstvennyy universitet, Khimicheskiy fakul tet) TITLE: Reaction of manganese and scandium oxides in air TOPIC TAGS: manganese compound, scandium compound, chemical reaction AN SSSR. Izvestiya. Neorganicheskiye materialy, v.2, no.2, 1966, SOURCE: 275-280 ABSTRACT: A table shows the composition of the samples investigated, the calcining temperature, and the calcining time. The mole % content of scandium oxide in the samples varied from 0 to 100%, the calcining temperature from 700 to 1100°C, and the calcining time from 2 to 100 hours. The starting samples were prepared by precipitation of scandium and manganese hydroxides by a mixture of NH40H + H202 from nitric acid solutions. The samples were calcined in a platinum boat at 700-150000 and then quenched in liquid nitrogen. An X-ray analysis was made of the samples. An NTR-62 unit was used for thermal analysis. The magnetic susceptibility was determined by the Faraday method. The article gives a phase diagram of the system, constructed from the experimental data. Card UDC: 546'713-31 + 546.631-31

# acc NR. AP6007251 established the existence of a compound with the composition ScMn03 and three types of cubic solid solutions; based on Sc203, Mn203, and a cubic solid solution Mn304. The compound ScMn03 crystallimodification with the composition Mn304. The compound ScMn03 crystallimodification vith the composition Mn304 to 2000 and the cubic modification of solid solutions based on Sc203 and the cubic modification formation of solid solutions based on Sc203 and the cubic modification Mn304. The solubility of Mn203 in scandium oxide changes only slightly formation of solid solutions based on Sc203 and the cubic based on Sc203 in with temperature and is from 17 to 20 mole %; the solubility of Sc203 in with temperature and is from 10.5 mole % at 1200°C to 30.0 mole % at 1500°C. The article demonstrates further that scandium oxide does not 1500°C. The article demonstrates further that scandium oxide does not form compounds or a wide range of solid solutions with Mn0, N10, Co0, cod0, and Zn0. Orig. art. has: 5 figures and 4 tables. SUB CODE: 07/ SUBM DATE: 30Jul65/ ORIG REF: 002/ OTH REF: 003

POLACZKOWA, Wanda; POROWSKA, Natalia; DYBOWSKA, Barbara

The influence of phenyl substituents in the benzene ring on the non-neighboring functional group. I. Benzoic acid derivatives. Rocz chemii 35 no.5:1263-1271 '61.

1. Department of Organic Chemistry, Institute of Technology, Warsaw and Institute of Organic Synthesis, Polish Academy of Sciences, Warsaw.

APPROVED FOR RELEASE: 06/15/2000 CIA-RDP86-00513R001342520013-0"

POLAND / Organic Chemistry. Synthetic Organic Chemistry.

G-2

Abs Jour: Ref Zhur-Khimiya, 1958, No 17, 57381.

Author Porowske News

Inst : Not given.
Title : Derivatives of the 1,2,3-Trimethylbenzene.

Orig Pub: Roczn. chem., 1957, 31, No 2, 677-679.

Abstract: 2.33 gr of 2,6-bis-(methyl chloride)-4-nitrotoluene (I) in 100 cc of alcohol is hydrated over
0.04 gr Pt02 yielding 90% of 3,4,5-trimethylaniline
(II). A solution containing 19.7 gr I in 50 cc
(H3COOH is heated for 5 hours with 70 gr Sn and
300 cc HCl yielding 96% II of 75-76.50 melting

card 1/2

POLACZKOWA, Wanda; POROWSKA, Natalia; DYBOWSKA, Barbara

Studies on the influence of phenyl substituents in the benzene ring upon the not neighboring functional group. Rocz chemii 36 no.1:41-50 \*62.

1. Department of Organic Chemistry, Institute of Technology, Warsaw Laboratory No.4, Institute of Organic Synthesis, Polish Academy of Sciences, Warsaw.

POROMSKA, N.

SCIENCE

PERIODICAL: ROCZNIKI CHEMII, Vol. 31, No. 2, 1957

POROWSKA, N. Derivatives of 1, 2, 3,-trimethylbenzene. p 677.

Monthly List of East European Accessions (EEAI) LC Vol. 8, No. 4
April 1959, Unclass

Porowska N.

Porowska N. "3-Indol-Butyric Acid." (Kwas 3-inkolilomaslowy). Przemysl Chemiczny, No 6, 1950, pp. 340-343.

In order to secure a higher yield and degree of purity - certain modifications were introduced into Jackson and Manske's method of 3-indol-butyric acid preparation. The reaction of obtaining phemylhydrazone deritative of an acidic ester of 2-ketopimelic acid from cyclohexanone-2-carboxylate and benzene diazonium chloride,--was carried out in a dilute alkaline solution, without excess of KOH, and resulted in producing phenyl-hydrazone of 94% yield without pigment adixture. The decarboxylation process of indol-2-carbozy-3-butyric acid was performed under diminished pressure (a-bout 30 mm). The conditions described make possible production of 3-indol-butyric acid acid from cyclohexanene reaches 22% of the theoretical amount.

SO: Polish Technical Abstracts No. 2, 1951

E-2

POROWSKA, N.

Poland/ Organic Chemistry - Synthetic organic chemistry

Abs Jour : Referat Zhur - Khimiya, No 4, 1957, 11698

: Butesin Picrate (Picrate of n-Butyl Ester of p-Aminobenzoic Acid) Author

Orig Pub : Pikrynian butezyny (pikrynian p-aminobenzoesanu n-butylu).

Przem. chem., 1953, 9, No 12, 624-625 (Polish)

There has been synthesized the picrate of p-NH2C6H5COCO4H9 (I, II--Abstract :

ester) -- the medicinal preparation "Butesin". 0.1 mole p-NH $_2$ C $_6$ H $_5$ -

COOH esterified by boiling (6 hours) with 0.66 mole  $n\text{-}C_{14}H_{\mbox{\scriptsize Q}}\mbox{OH}$  and

0.115 mole 92% H<sub>2</sub>SO<sub>lt</sub>, added 100 ml water, steam distilled, recove-

ring 30 g  $C_{l_1}H_{Q}^{\ OH}$ , residue cooled to  $0^{\circ}$ , thus obtaining the sulfate of

II, MP 184-189°; latter is dissolved in hot water and  $\sim$  15 mL of

concentrated NH4OH are added, yield of II 70-72%, BP 173-1740/8 mm,

Card 1/2

APPROVED FOR RELEASE: 06/15/2000 CIA-RDP86-00513R001342520013-0"

Poland/ Organic Chemistry - Synthetic organic chemistry

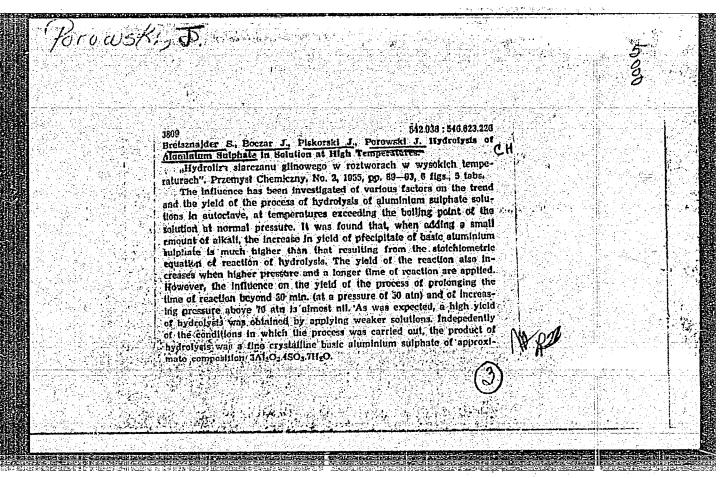
Abs Jour : Referat Zhur - Khimiya, No 4, 1957, 11698

E-2

MP 58-59°. II is also prepared by reduction (Fe + HCl, 90°) of p-NO<sub>2</sub>C<sub>6</sub>H<sub>5</sub>COOC<sub>4</sub>H<sub>9</sub>, yield 70-72%. Solution of 0.025 mole of picric acid and 0.05 mole II, in 100 ml alcohol, is boiled 0.5 hour, cooled, poured dropwise into 700 ml water, to obtain I, yield 95%, MP 110-111.5°.

Card 2/2

Porowska./I			
Derivatives of 1, rowska (Politechnika 077-9(1957)(English s b <sub>16</sub> 140-1°, and Me es 40.5-1.5°, 3,4,6-Me <sub>3</sub> C <sub>4</sub> H <sub>2</sub> Br, b <sub>17</sub> 117-18°	2.3-trimethylbenzens. Nata . Warsaw). Rossniff Che uimmary).—The nitrile, m. 96 ter of 3.4,5-trimethylbenzoic s .H.Cl. by 105-8°, and 3.4 were described. A. Kregle	IIA P6- m. 31: 5-7.5° 44 50 etd. m. 5-Mc- wald 1/46 5	oeg) indig



KOPCZYNSKI, Cezary, inz.; POROWSKI, Jozef, mgr inz.

Induction hardening of big crankshafts. Przegl mech 23 no. 3: 78-81 10 F 164.

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	Kuczewshi S., Porowski L. Characteristics of the I.T.C. Research Bay for Examining Iligic Busic Systems, and Notes on the Test Methods and Measuring Instruments Employed.  "Charakterystyka stolska ITC do badan pallsad nieruchomych I uwagi na temat zastosowanych metod badawczych i przyrządów pomiarowych. (Prace Gł. Inst. Mechah. No. 3). Warszewa 1951, PWT, 8 pp., 21 ligs.  Research over the flow gases through the blade system. Other research bays. Research bay at the Institute for Thermal Engineering. Examination of the layer next to the wall, Influence of gaps at the apex of blades. The testing of blades of varying height, Rates of flow. Models of blade systems. Design and operation of the research bay. Measuring technique adopted. Measuring instruments.		
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Parawski, b	Kuczewski S., Porowski L. Determination of Polential Flow Losses in Cascade and Blade Passages.  "Określenie strat przepływów przez polisady i wieńce kierownicze", (Prace Inst. Mechan. No. 6), Warszawo, 1953, PWT, 4 pp., 4 figs.  The authors review the possibilities for measuring, on the basis of measurement of total pressures, the coefficient of losses in the flow through stationary blades. They consider errors occurring in two instances — of blades with kinematic reaction stages of 0.5 and 1.0. The authors' method makes it possible to determine, with a high degree of accuracy, the aerodynamical coefficients for an instance of stationary blades with a kinematic reaction stage approaching 1.0.				
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KLOSOWSKI, Andrzej; POROWSKI, Ludwik

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1. Biprohut, Warsaw Branch.

KLOSOWSKI, Andrzej; POROWSKI, Ludwik

Thermal calculations of regenerators by using electronic computers. Problemy proj hut maszyn 13 no.3:77-80 ½ '65.

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POROWSKI, Stanislaw; LOPATKIEWICZ, Janina

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(APHASIA)

(HEARING)

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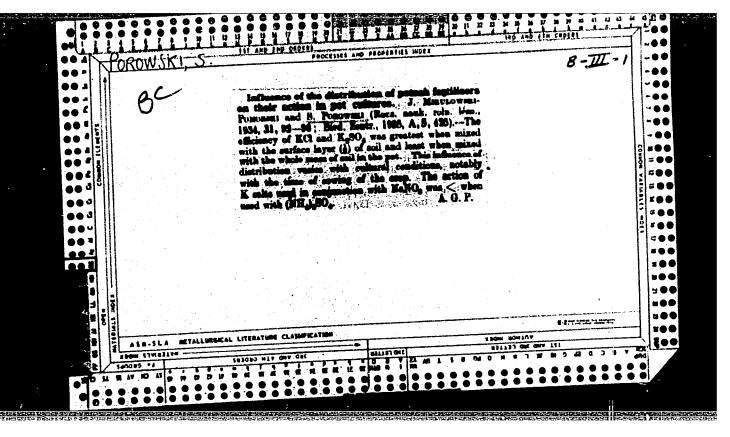
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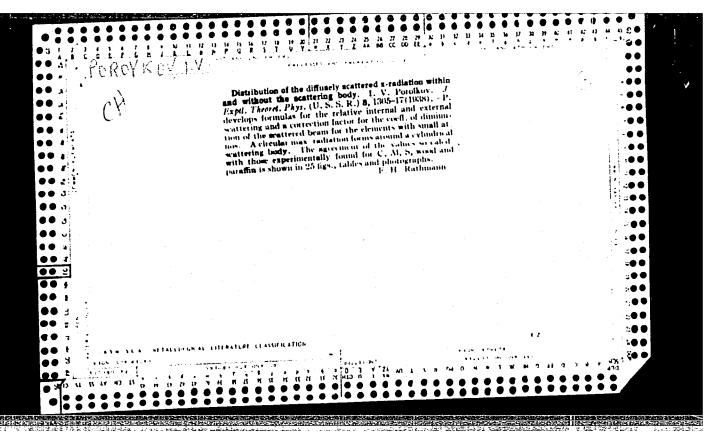
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	Normalizac	ja P 30 n	0.1:9-1	2 '62.					
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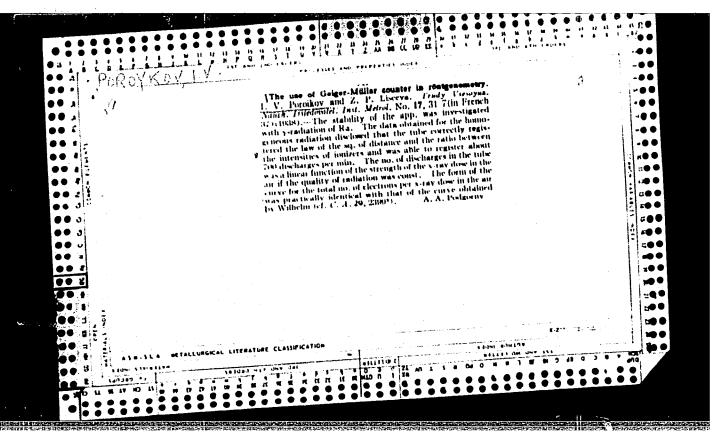


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TREASURE ISLAND BIBLIOGRAPHICAL REPORT

AID 678 - I

PHASE I

call No.: QC481.P67

BOOK

Author: POROYKOV, I. V. Full Title: X-RAY MEASUREMENT

Transliterated Title: Rentgenometriya

PUBLISHING DATA

Publishing House: State Publishing House of Technical and Theoretical

Literature ("Gostekhizdat")

No. pp.: 383 Date: 1950

No. of copies: 3,000

Contributors: Gusev, N. G., Krongauz, A. N., Poroykova, A. V. PURPOSE: The book is intended for engineers and scientific workers in

different fields and physicians whose activities require the use of

X-rays.

Coverage: This book discusses physical and technical fundamentals and TEXT DATA practical methods of X-ray measuring in roentgen units. In part-I (p. 9-151) attention is given to problems of nuclear physics, absorption and scattering, and to the secondary processes caused by X-rays. It deals also with the metrology of Roentgen radiation and the dosimetric characteristics of radiation sources, as well as with radiation spectra. —Part II, "Practical Roentgenometry" (p. 152-257), radiation spectra. —easuring instruments, problems of the biological discusses radiation—measuring instruments, problems of the biological

Rentgenometriya

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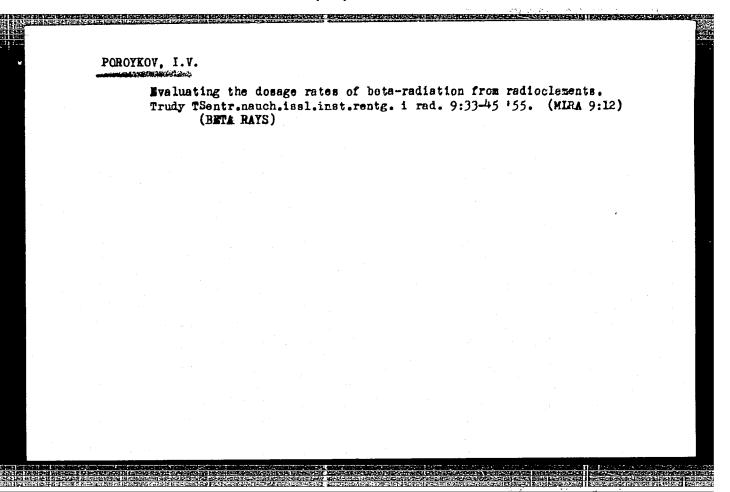
effects of radiation, and methods and means of radiation shielding. Shielding materials are examined as well as control measures for roentgenometers and shields. Methods of using X-rays for purposes of analysis are also discussed. At the end, are five supplements-(p. 258-380) containing various tables. Other tables, many illustrations, diagrams and equations are profusely scattered throughout the book. According to the author, Soviet science has a leading role in the development of X-ray measuring methods. Some references are also made to gamma-ray measurement.

No. of References: Part I, 17 refs., 5 Russian (1934-1949); Part II,

12 refs., 8 Russian (1939-1949).

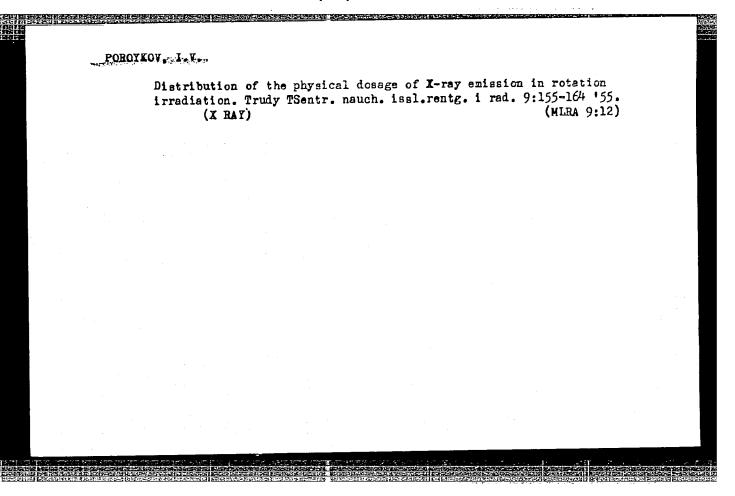
Facilities: All-Union Institute of Metrology im. D. I. Mendeleyev; State Radium Institute (GRI); Kiyev Radium Institute (KRI); A. I. Tkhorzhevskiy, K. K. Aglintsev, Ya. L. Shekhtman, A. K. Trapeznikov, V. V. Dmokhovskiy, A. I. Ruderman and others.

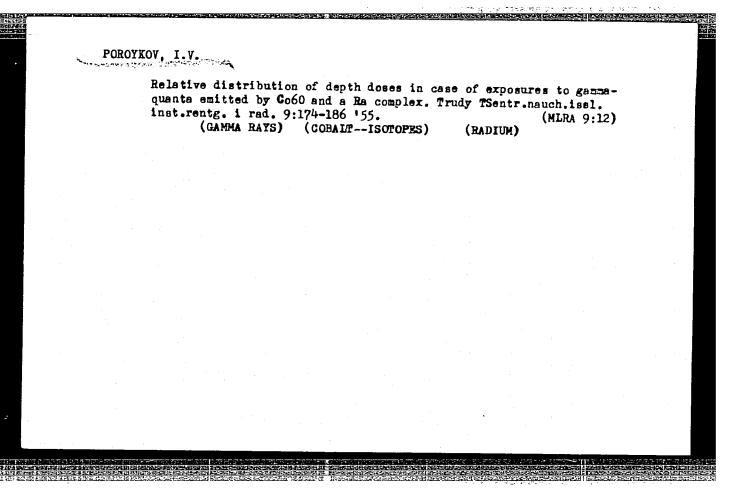
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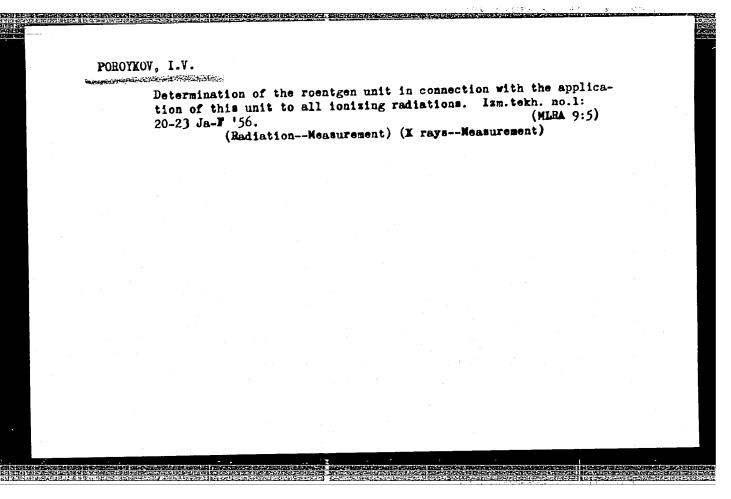


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[Radiography of castings] Rentgenografiia otlivok. Moskva, Mashinostroenie, 1965. 95 p. (MIRA 18:3)







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[Selected chapters on physics] Guiel'nye glavy flaiki.

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IVANOV, Nikolay Nikolayevich, prof.; POROYKOV. Ivan Vasil'yevich, prof.; FIRSTOV, V.G., red.; ZUBKOVA, M.S., red.izd-va; DONSKAYA, G.D., tekhn.red.

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(MIRA 12:1)

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MALYSHEV, Yu.M., kand. ekonom. nauk, otv. red.; SHMATOV, V.F., kand. ekonom. nauk, otv. red.; POROYKOV, Yu.D., red.; SHAFIN, I.G., tekhn. red.

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YAKHIMOVICH, Varvara L'vovna; CLLI, A.I., prof., doktor geol.-mineral. nauk, otv.red.; POROYKOV, Yu.D., red.; KOBYAKOV, I.A., tekhn.red.

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[Geomorphology and recent tectonics of the Volga-Ural region and the Southern Urals] Geomorfologia i noveishaia tektonika Volgo-Ural'skoi oblasti i IUzhnogo Urala; trudy. Ufa. Akad.nauk SSSR, Bashkirskii filial, Gorno-geol.in-t, 1960. 347 p.

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YAKHIMOVICH, Varvara L'vovna; OLLI, A.I., prof., doktor geologo-mineralog. nauk, otv.red.; POROYKOV, Yu.D., red.; KALAGANOV, I.S., tekhn.red.

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KONAREV, V.I., prof., otv.red.; BELOZERSKIY, A.N., red.; GENKEL¹, P.A., prof., red.; SERGEYEV, L.I., prof., red.; MAZILKIN, I.A., kand. biolog.nauk, red.; KHANISLAMOV, M.G., kand.sel¹skokhoz.nauk, red.; POROYKOV, Yu.D., red.; VALEYEV, G.G., tekhn.red.

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POROYKOVA, A.I.; NALBANDYAN, A.B.

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1. Institut khimicheskoy fiziki AN SSSR. Submitted March 20, 1965.

ORG: Institute of Chemical Physics, AN SSSR (Institut khimicheskoy fiziki AN SSSR)  TITLE: Oxidation mechanism of propane in the presence of hydrogen bromide and bromine. I. The reaction of propyl and isopropyl hydroperoxides with hydrogen bromia and bromine in the gas phase  SOURCE: Armyanskiy khimicheskiy zhurnal, v. 19, no. 2, 1966, 83-88  TOPIC TAGS: hydrocarbon oxidation, reaction mechanism, combustion  ABSTRACT: It was found that isopropyl hydroperoxide and n-propyl hydroperoxide react with HBr and Br <sub>2</sub> at room temperature to form, respectively, acetone and a mixture of propionaldehyde and n-propyl alcohol. The reaction between isopropyl hydroperoxide and HBr follows second-order kinetics; the reaction rate constant in the range 18—62C is  The investigation of the reaction of hydroperoxides with Br <sub>2</sub> is complicated by the	ACC NRI AP60	14401	SOURCE CODE: UR	/0426/66/019/002/008:	3/0088
TITLE: Oxidation mechanism of propane in the presence of hydrogen bromide and bromine. 1. The reaction of propyl and isopropyl hydroperoxides with hydrogen bromiand bromine in the gas phase  SOURCE: Armyanskiy khimicheskiy zhurnal, v. 19, no. 2, 1966, 83-88  TOPIC TAGS: hydrocarbon oxidation, reaction mechanism, combustion  ABSTRACT: It was found that isopropyl hydroperoxide and n-propyl hydroperoxide react with HBr and Br <sub>2</sub> at room temperature to form, respectively, acetone and a mixture of propionaldehyde and n-propyl alcohol. The reaction between isopropyl hydroperoxide and HBr follows second-order kinetics; the reaction rate constant in the range 18—62C is  A <sub>i</sub> = 0.8·10 <sup>-11</sup> e	AUTHOR: Por	oykova, A. I.; Voyevod	skiy, V. V.; Nalbandya	n, A. B.	47
bromine. I. The reaction of propyl and isopropyl hydroperoxides with hydrogen bromi and bromine in the gas phase  SOURCE: Armyanskiy khimicheskiy zhurnal, v. 19, no. 2, 1966, 83-88  TOPIC TAGS: hydrocarbon oxidation, reaction mechanism, combustion  ABSTRACT: It was found that isopropyl hydroperoxide and n-propyl hydroperoxide react with HBr and Br <sub>2</sub> at room temperature to form, respectively, acetone and a mixture of propional dehyde and n-propyl alcohol. The reaction between isopropyl hydroperoxide and HBr follows second-order kinetics; the reaction rate constant in the range 18—62C is    Mai = 0.8 · 10 <sup>-11</sup> e   Cm <sup>3</sup> molecule sec	ORG: Instit	ute of Chemical Physic	s, AN SSSR (Institut kh	imicheskoy fiziki AN	SSSR)
TOPIC TAGS: hydrocarbon oxidation, reaction mechanism, combustion  ABSTRACT: It was found that isopropyl hydroperoxide and n-propyl hydroperoxide react with HBr and Br <sub>2</sub> at room temperature to form, respectively, acetone and a mixture of propional dehyde and n-propyl alcohol. The reaction between isopropyl hydroperoxide and HBr follows second-order kinetics; the reaction rate constant in the range 18—62C is $\frac{9200\pm1500}{RI}  \text{cm}^3$ moleculæ sec	bromine. T.	the reaction of propyl	pane in the presence o and isopropyl hydrope	f hydrogen bromide ar roxides with hydroger	id i bromi
ABSTRACT: It was found that isopropyl hydroperoxide and n-propyl hydroperoxide react with HBr and Br <sub>2</sub> at room temperature to form, respectively, acetone and a mixture of propional dehyde and n-propyl alcohol. The reaction between isopropyl hydroperoxide and HBr follows second-order kinetics; the reaction rate constant in the range 18—62C is $\frac{9200\pm1500}{RT} \frac{\text{cm}^3}{\text{molecule-sec}}$	SOURCE: Arm	yanskiy khimicheskiy z	hurnal, v. 19, no. 2,	1966, 83-88	
react with HBr and Br <sub>2</sub> at room temperature to form, respectively, acetone and a mixture of propional dehyde and n-propyl alcohol. The reaction between isopropyl hydroperoxide and HBr follows second-order kinetics; the reaction rate constant in the range 18—62C is $\frac{9200\pm1500}{RI}  \text{cm}^3$ $k_1 = 0.8 \cdot 10^{-11}  e^{-\frac{9200\pm1500}{RI}}  \text{cm}^3$	TOPIC TAGS:	hydrocarbon oxidation	, reaction mechanism,	combustion	
$k_1 = 0.8 \cdot 10^{-11} e \frac{RT}{\text{molecule} \cdot \sec \cdot}$	react with H ture of prop peroxide and	Br and Br <sub>2</sub> at room tem Lonaldehyde and n-prop HBr follows second-or	perature to form, respe yl alcohol. The react	ctively, acetone and ion between isopropy	a mix- l hydro
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	The investig	ation of the reaction	of hydroperoxides with	Br <sub>2</sub> is complicated b	y the

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fact that Orig. art.	the real	action take 5 figures.	s place at	the wall,	as well	as in th	e homogened	ous phase [VS]
SUB CODE:	21/	SUBM DATE:	01Sep65/	ORIG REF	012/ 0	TH REF:	003/ ATD	PRESS 42
Card 2/2								

L:26001-66 EWT(m)/EWP(j)/T DS/WE/RM ACC NR: AP6015616 SOURCE CODE: UR/0020/66/168/002/0386/0387 AUTHOR: Poroykova, A. I.; Voyevodskiy, V. V. (Academician); Nalbandyan, (Academician AN ArmSSR) ORG: Institute of Chemical Physics, Academy of Sciences SSSR (Institut khimicheskoy fiziki Akademii nauk SSSR) TITLE: Quantum yield of acetone and length of the reaction chain in photochemical oxidation of propane in the presence of bromine SOURCE: AN SSSR. Doklady, v. 168, no. 2, 1966, 386-387 TOPIC TAGS: photochemical oxidation, propane, acetone, quantum yield, chain reaction kinetics ABSTRACT: Bromine-initiated photochemical oxidation of propane has been studied to determine quantum yield & of the main product, acetone, and the length of the reaction chain,  $v = 1/2\Phi$ . Oxidation was carried out in a jet vacuum apparatus at 202C with illumination by a mercury quartz lamp under given partial pressures of reactants. The degree of conversion was small and the rate of acetone formation constant. The formula was used to calculate quantum yield, where  $I_0$  is the rate of initiation and

rate of HB	0015616  )/dt is the rat r formation in ction was carri rmation were ca of acetone was	the photocher ed out in th lculated fro	nical reactions same apparate the time-de	n of hydroge tus at 266C. pendence of	The rates of the acetone	of acetone and HBr
	-07/ SUBM DATE	: 17Aug65/	ORIG REF: 0	06/ OTH REF	: 006/ ATD	PRESS:4255
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ACC NR: AP6	014402 SOURCE CODE: UR/0426/66/019/002/0089/0095
AUTHOR: Po	proykova, A. I.; Voyevodskiy, V. V.; Nalbandyan, A. B.
Aumor.	Burney Constitution of the
ORG: Insti	tute of Chemical Physics, AN SSSR (Institut khimicheskoy fiziki AN SSSR)
TITLE: Qxi	dation mechanism of propane in the presence of hydrogen bromide and
	. Evaluation of the rate constant of the reaction of the RO2 radical with
hydrogen br	'omide' '가도 말을 다 있다. 그리고 있는 그는 그는 그를 들었다. 그런 가는 그리고 말을 다 했다.
SOURCE: Ar	myanskiy khimicheskiy zhurnal, v. 19, no. 2, 1966, 89-95
TOPIC TAGS:	hydrocarbon oxidation, reaction mechanism, propulsion
ABSTRACT:	The photochemical oxidation of propane, initiated by addition of small hydrogen bromide (0.5—5%), was studied in the temperature range 150—2400
Acetone was	found to be the main product of oxidation. The small amounts of acetal-
dehyde form	med in the course of the reaction easily undergo further reactions under
to measure	mental conditions employed. The fast stream method used made it possible the amounts of acetaldehyde formed and to determine the ratio of the re-
	constants for reactions (1) and (2):
	$(CH_3)_2CHO_2 + HBr \longrightarrow (CH_3)_2CHO_3H + Br \longrightarrow CH_3COCH_3 + Br + H_3O, (1)$
	CH <sub>3</sub> COCH <sub>3</sub> +Br + H <sub>3</sub> O
	(сн,),сно, → сн,сно + сн,о. (2)
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L 23823-6 ACC NR: AP The ratio	6014402	n rate constar	nts was found to	o be		•
		$\frac{K_1}{K_2} = 10^{-22.9} e$				[Vs
	has: 4 figur 21/ SUBM DAT		ORIG REF: 015	/ oth ref:	004/ ATD PR	
						1.56

ACC NR: AP6014403	SOURCE CODE: UR/0426/66/019/002/0096/0110
AUTHOR: Poroykova,	A. I.; Voyevodskiy, V. V.; Nalbandyan, A. B.
ORG: Institute of	Chemical Physics, AN SSSR (Institut khimicheskoy fiziki AN SSSR)
TITLE: The mechani bromine. III. Photo	sm of propane oxidation in the presence of hydrogen bromide and chemical oxidation of propane in the presence of Br2
SOURCE: Armyanskiy	khimicheskiy zhurnal, v. 19, no. 2, 1966, 96-110
TOPIC TAGS: hydroc	arbon oxidation, reaction mechanism, combustion
	레이트 네이트를 들는 학교가는 이 하면 되고 이 물론이라면 하는 수 있다. 그는 그는 그 전에 가는 그 없다.
studied in the tempowhen HBr is present nant reaction production and well as to	ochemical oxidation of propane, initiated by addition of Br <sub>2</sub> , was erature range 160—240C. In the early phase of the reaction, only in negligible amounts, acetone was found to be the predomict. This observation, together with a number of kinetic relation he absence of CH <sub>2</sub> OH in the reaction products, indicate that the
studied in the tempowhen HBr is present nant reaction production and well as to	ochemical oxidation of propane, initiated by addition of Br2, was erature range 160—240C. In the early phase of the reaction, only in negligible amounts, acetone was found to be the predomict. This observation, together with a number of kinetic relation
studied in the tempowhen HBr is present nant reaction production and well as to	ochemical oxidation of propane, initiated by addition of Br <sub>2</sub> , was erature range 160—240C. In the early phase of the reaction, only in negligible amounts, acetone was found to be the predomict. This observation, together with a number of kinetic relation he absence of CH <sub>3</sub> OH in the reaction products, indicate that the e can best be explained by the following reactions alone:

1	NR: AP	6U144	U3												(
The	possib.	le re	action	iso.C <sub>1</sub>	H,O, + Br	, <del></del> -	CH,C	OCH +	- HBc-l	- BrO	and t	he med	han1	sm of	
acei	cone and	a HBT	torma	lon, w	mich is	in go	od ag	reeme	ent w	1th	the ex	perime	nta1	data,	
exan	nined.	Orig	. art.	has:	6 figu	res.									[VS
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5 (4), 5 (3)

AUTHORS:

Porbykova, A. I., Voyevodskiy, V. V., SOV/76-33-6-26/44

-Nalbandyan, A. B.

TITLE:

Photoinitiation of Propage Oxidation in the Pressure of

Ammenia and Hydrogen Sulphide (Foteinitaiirovanija okisleniya

propana v prisutstvii ammiaka i serovedoreda)

PERIODICAL:

Zhurnal fluicheskoy khimii, 1959, Vol 35, Nr 6,

pp 1336-1344 (USSR)

ABSTRACT:

The thermal exidation of low, gaseous paraffin hydrocarbons proceeds only at high temperatures at a noticeable rate; the high temperature leads to a decay of the intermeliate products so that these products as well as the reaction kinetics cannot

be investigated under these conditions. A photochemical meastion initiation (RI) is, besides the catalysis, an important method of (RI). The photochemical exidation of low

gaseous hydrocarbons was first investigated by A. B. Nalbardyan et al (Refs 1-3), and among other things, a reaction mechanism of the propage oxidation at low temperature was suggested (i) - (8). The photolyses (P) of

the ammonia (I) and hydrogen sulphide (II) have been

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insufficiently examined up to new; on the other hand, it

Photoinitiation of Propane Oxidation in the Presence SOV/76-33-6-26/44 of Ammonia and Hydrogen Sulphide

must be assumed that at a (P) of (I) or (II) in the presence of a mixture of propane (III) and oxygen (IV), the resulting inorganic radicals will form propyl radicals with the molecules of (III), thus initiating the (III)-oxidation. The latter has already been observed (Refs 16, 17), the mode of origin of the formed acetone could not be clarified. The present experiments were carried out in a vacuum device (Fig 1), which was equipped with 2 quartz lamps PRK-2. The reaction products were frozen out by liquid nitrogen. The experiments led to the following statements: Isopropylhydroperoxide (V), acetaldehyde and formaldehyde form at 200-220°C as main products of the photochemical propane oxidation. The (RI) with (I) gives at 220°C a yield of reaction products of 5 per decomposed (I)-molecule, which is considered as a confirmation of the chain mechanism in the process. The present results as well as those obtained by A. B. Nalbandyan et al permit the assumption that the propane oxidation, photoinitiated with mercury (Refs 1-3), ammonia or hydrogen sulphide, proceeds according to the same chain mechanism, independent of the type of initiator. The acetone which - as

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Photoinitiation of Propane Oxidation in the Presence SOV/76-33-6-26/44 of Ammonia and Hydrogen Sulphide

mentioned above ~ is observed in static experiments, is produced by the decay of (V). There are 5 figures, 1 table,

and 23 references, 7 of which are Soviet.

ASSOCIATION: Akademiya nauk SSSR, Institut khimicheskoy fiziki Moskva

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SUBMITTED: November 27, 1957

Card 3/3

POROYKOVA, G.D.

Circulatory modifications in cardiovascular diseases caused by physical stress. Klin. med., Moskva 29 no.12:82-83 Dec 51. (CIML 21:4)

1. Of the Fourth Moscow Municipal Clinical Hospital (Head Physician V.A. Tveritin) and of the Propedeutic Therapeutic Clinic of the Therapeutic Faculty (Director--Prof. V.M. Volin), Second Moscow Medical Institute imeni I.V. Stalin.

LEVIE, A.Ye.; POROYKOVA, G.D.

Treatment of gastric and duodenal ulcer with hyaluronic acid. Sovet. med. 16 no.4:21-22 Apr 1952. (CLML 22:1)

1. Of the Department of the Propedeutics of Internal Diseases (Head -- Prof. M. A. Volin), Second Moscow Medical Institute imeni I. V. Stalin, and of Fourth Moscow Municipal Clinical Hospital (Head Physician -- P. G. Demidov).

POROYKOVA, G. D.

"Physical Stress As a Method of Determining the Functional Condition of the Cardiovascular System During Blood Circulation Insufficiency." Canadided Sci, Second Moscow Redical Inst, Moscow, 175h. (RZhBiol, No 3, Feb 55)

SO: Sum. No. 631, 26 Aug 55 - Survey of Scientific and Technical Dissertation Defended at USSR Higher Educational Institutions. (14)

POROIKOVA, G.D., kandidat meditsinskikh nauk

Experience in the use of therapeutic physical education in compound therapy of insufficiency of blood circulation. Sov.med. 19 no.4:53-62 Ap '55.

1. Iz IM Moskovskogo meditsinskogo instituta imeni I.V.Stalina.
(BLOOD CIRCULATION, dis., insuff., ther., physiother.)
(PHYSICAL THERAPT, in various dis., insuff. of blood circ.)

# POROYKOVA, G.D., kandidat meditsinskikh nauk THE PROPERTY OF THE PARTY OF TH Effect of some physical stresses and respiration on venous pressure in circulatory insufficiency. Sov.med. 19 no.6:41-44 Je '55. (MLRA 8:9) 1. Is kafedry fizicheskogo vospitaniya i vrachebnoy fizicheskoy kul'tury (sav. S.M. Ivanov) II Moskovskogo meditsinskogo instituta imeni I.V. Stalina. (EXERCISE, effects. on venous pressure in cardiovasc. insuff.) (RESPIRATION. eff. of various methods of resp. on venous pressure in cardiovasc. insuff.) (BLOOD PRESSURE, eff. of exercise & various methods of resp. on venous pressure in cardiovasc. insuff.) (CARDIOVASCULAR DISEASES, physiology. eff. of exercise & various methods of resp. on venous pressure in cardiovasc. insuff.)

POROYKOVA, G.D.

Effect of physical stress on the oxygen saturation of arterial blood in patients with circulatory insufficiency. Sov. med. 24 no. 10:33-40 0 '60. (MIRA 13:12)

l. Iz gospital'noy terapevticheskoy kliniki (dir. - prof. P.Ye.
Lukomskiy) II Moskovskogo meditsinskogo institaimeni N.I. Pirogova).
(EXERCISE) (HEART FAILURE)
(BLOOD---OXYDEN CONTENT)

POROYKOVA, L.N., mladshiy nauchnyy sotrudnik

Qualitative reactions to some sulfanilamide preparations. Sior. nauch. trud. TSANII 3:120-124 62. (MIRA 16:11)

l. Laboratoriya farmatsevticheskogo analiza (rukovoditel laboratorii - dotsent, kand.khim.nauk M.I.Tarasenko) TSentral nogo aptechnogo nauchno-issledovatel skogo instituta.

YEREMEYEVA, V.S.; POROYKOVA, L.N.; PETROVA, R.I.; MORUNOVA, Z.S.; SIVITSKAYA, O.K.

Use of an internal indicator in the nitritometric titration of drugs.

Apt. delo 9 no.3:60-63 My-Je '60. (MIRA 14:3)

(DRUGS) (COLORIMETRY)

(INDICATORS AND TEST-PAPERS)